



Pergamon

Tetrahedron Letters 41 (2000) 8285–8288

TETRAHEDRON  
LETTERS

# Intramolecular Mitsunobu reaction in the regio- and stereoselective synthesis of *cis*-4,5-disubstituted piperidin-2-ones

Nicole Langlois\* and Olivier Calvez

*Institut de Chimie des Substances Naturelles, CNRS, 91198 Gif-sur-Yvette, France*

Received 17 July 2000; accepted 30 August 2000

---

## Abstract

Enantiopure *cis*-4,5-disubstituted piperidin-2-ones were synthesized through the cyclization of *O*-benzyl hydroxamates under Mitsunobu conditions, followed by samarium diiodide reduction. © 2000 Elsevier Science Ltd. All rights reserved.

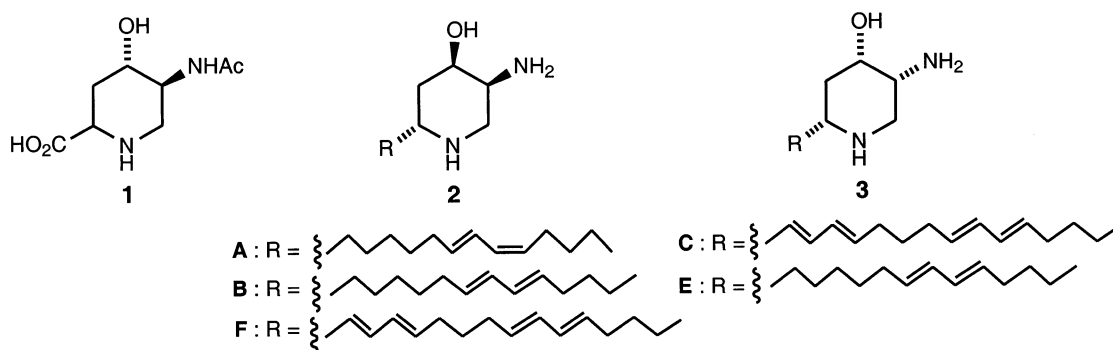
*Keywords:* polysubstituted piperidinones; hydroxamates; Mitsunobu reaction; samarium diiodide.

---

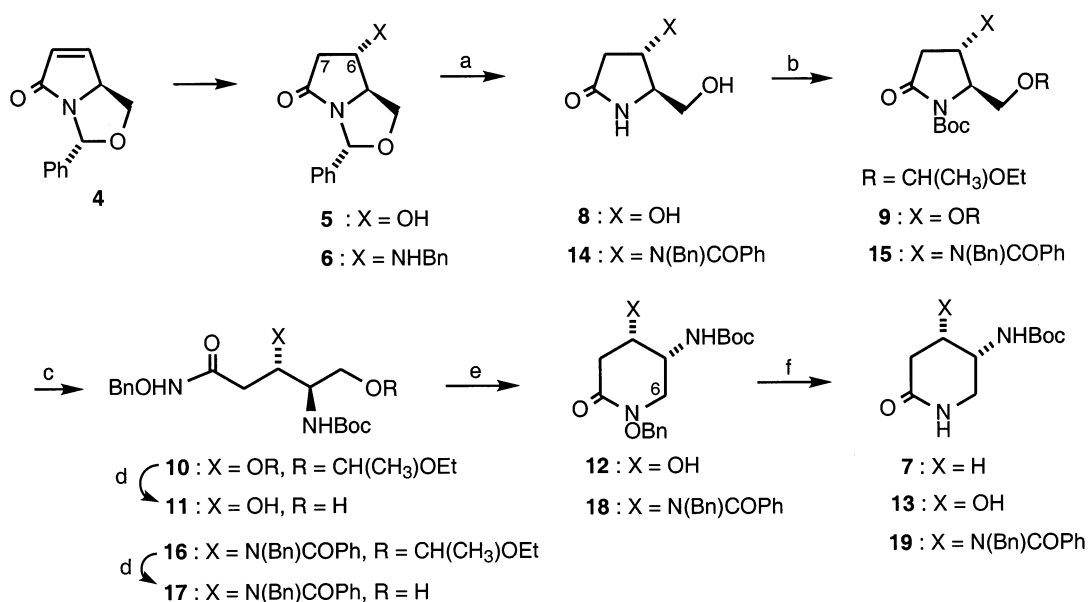
Polysubstituted piperidines are common structural sub-units in numerous natural products and drugs. As this class of compounds exhibit a wide range of biological activities,<sup>1</sup> regio- and stereoselective syntheses of enantiopure substituted piperidines are a target of particular interest.<sup>2</sup> Thus, many structural analogues of pyranose-type azasugars, particularly 2-substituted 3,4,5-trihydropiperidines with transition-state like half-chair conformation, have been synthesized and studied as glycosidase inhibitors.<sup>3,4</sup> Among the 2,4,5-trisubstituted members, (4*S*,5*S*)-5-acetamido-4-hydroxypiperidic acids **1** are weak sialidase inhibitors,<sup>5</sup> whereas pseudodistomins A, B, F (**2**) and E, C (**3**) are known for their activity as calmodulin antagonists and/or their cytotoxicity against L1210 leukemia cells.<sup>6,7</sup> In these marine piperidines isolated from tunicates, a *cis* relationship is observed between the substituents at C-4 and C-5.<sup>6,7</sup> Focusing our interest on this characteristic structural feature, we report here a new route to the diastereo- and enantioselective construction of *cis*-4,5-disubstituted piperidin-2-ones, the amide carbonyl of which allows further functionalization.

---

\* Corresponding author. Fax: 0169077247; e-mail: nicole.langlois@icsn.cnrs-gif.fr

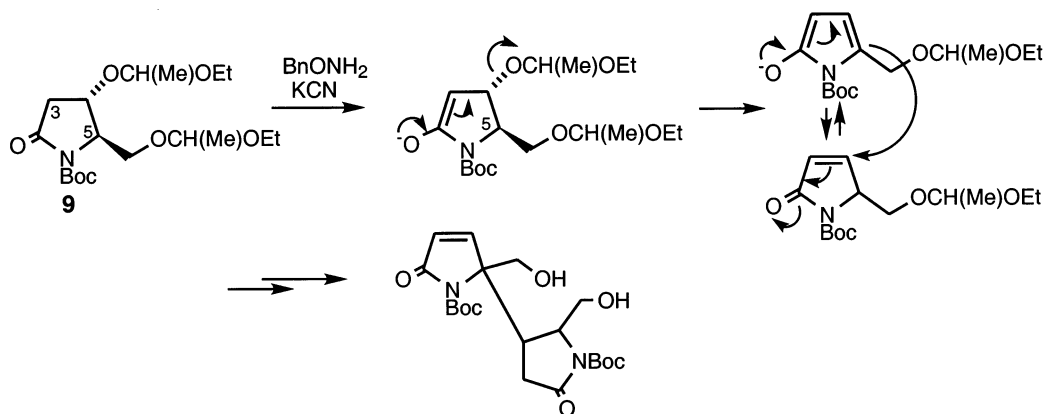


Starting from the rigid bicyclic  $\alpha,\beta$ -unsaturated lactam **4** derived from inexpensive (*S*)-pyroglutaminol, the synthetic route, summarized in the general Scheme 1, took advantage of the high diastereoselectivity observed in the substitution at C-6. The 6*S*-hydroxy derivative **5** was obtained by regiospecific opening of 6*R*,7*R*-epoxide with  $\text{SmI}_2$  at  $-78^\circ\text{C}$  (83% for two steps),<sup>8</sup> whereas the 6*S*-*N*-benzylamino compound **6** resulted from stereospecific conjugate addition of *N*-benzylamine (88%).<sup>9</sup> The nitrogen of the targeted piperidin-2-ones could be introduced by the opening of *N*-*tert*-butoxycarbonylpyrrolidin-2-one rings with *O*-benzylhydroxylamine to form hydroxamates. The low  $\text{p}K$  of *O*-benzyl hydroxamates, indeed, is known to induce a good chemoselectivity towards an intramolecular Mitsunobu reaction of substrates also containing carbamate functions. This selectivity has been widely used in  $\beta$  and  $\omega$ -lactam formation,<sup>10,11</sup> and has been extended in our laboratory to  $\delta$ -lactam preparation, as exemplified by the synthesis of enantiopure (*S*)-5-aminopiperidin-2-one **7**.<sup>12</sup>



Scheme 1. Reagents and conditions (all the reactions were performed at room temperature): (a)  $\text{CF}_3\text{CO}_2\text{H}$ ,  $\text{H}_2\text{O}$ -THF 1:1; (b) (i) ethylvinylether,  $\text{CCl}_3\text{CO}_2\text{H}$ ,  $\text{CH}_2\text{Cl}_2$ ; (ii)  $(\text{Boc})_2\text{O}$ , DMAP,  $\text{CH}_3\text{CN}$ ; (c)  $\text{BnONH}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{Na}_2\text{CO}_3$ ; (d) 0.1N HCl; (e)  $\text{PPh}_3$ , DEAD, THF; (f)  $\text{SmI}_2$ , THF

Accordingly, the alcohol **5** gave rise quantitatively to the diol **8** after hydrolysis of the oxazolidine protection with aqueous trifluoroacetic acid in THF. The diol **8** was converted into suitably protected *N*-Boc pyrrolidin-2-one **9** by classical methods (89% for two steps, Scheme 1). At first, *O*-benzyl hydroxamate **10** was obtained from **9** (40%) by triisobutylaluminum mediated transamidation<sup>13</sup> with *O*-benzylhydroxylamine, but this reaction was unreliable and *N*-Boc deprotection was observed in some cases. The cleavage in the presence of a catalytic amount of potassium cyanide,<sup>14</sup> or using lithium amide as the nucleophile, led to a mixture of products resulting from deprotonation at C-3 with  $\beta$ -elimination. The major product resulted from Michael addition of an unsaturated lactam to another molecule, providing a dimeric compound characterized after deprotection of the hydroxyl groups (Scheme 2).<sup>15</sup> The opening of the  $\gamma$ -lactam ring with the amine in the presence of H<sub>2</sub>O and Na<sub>2</sub>CO<sub>3</sub> was preferred, following our recently developed method.<sup>14</sup> Under these conditions, however, the reaction with **9** is very slow and was stopped before completion to avoid a possible  $\beta$ -elimination of the functional group at C-4. Compound **10** was isolated in 60% yield, together with starting **9** (25%).



Scheme 2.

A selective Mitsunobu reaction with *O*-deprotected diol **11** (100% from **10**) was performed in THF at room temperature using 1.2 equiv. of reagents (PPh<sub>3</sub>, DEAD) and provided **12** (68%). Previous cyclization through selective Mitsunobu reaction with unprotected amino-polyols leading to five-membered rings has been described in pyridine as solvent.<sup>16</sup> In the NMR spectra of the cyclization product **12**, the chemical shifts of C-6-H<sub>2</sub> (<sup>13</sup>C:  $\delta$  = 49.8 ppm), consistent with a NCH<sub>2</sub> methylene, prove that *N*-alkylation occurred versus *O*-alkylation. The N–O bond of **13** was cleanly cleaved by SmI<sub>2</sub> in THF at room temperature,<sup>17,18</sup> and the required piperidinone **13** was obtained in 84% yield (Scheme 1).

A similar scheme was applied to the synthesis of the protected *cis*-4,5-diamino piperidin-2-one **19** as an analogue. The 6-*N*-benzylamino derivative **6** led, after *N*-benzoylation (PhCOCl, 2.4 equiv., CH<sub>2</sub>Cl<sub>2</sub>, Et<sub>3</sub>N, rt, 95%) and acid hydrolysis (100%), to the  $\gamma$ -lactam **14**, which was converted into *N*-Boc pyrrolidinone **15** (70% for two steps). The  $\gamma$ -lactam ring of **15** was smoothly opened with *O*-benzylhydroxylamine to afford the pentanamide **16** (72%) and unreacted **15** (11%). These compounds gave broad signals in <sup>1</sup>H NMR, probably owing to the presence of conformers. By intramolecular alkylation under Mitsunobu reaction conditions, the primary alcohol **17** (100% from **16**) gave rise efficiently to 4,5-disubstituted piperidin-2-one **18** (82%). This compound was treated with SmI<sub>2</sub> to afford the 4,5-disubstituted piperidin-2-one **19**

in good yield (84%). In this way, *cis* cyclic vicinal diamines in diastereomerically and enantiomerically pure form become accessible and could be useful as scaffolds to synthesize not only more complex molecules but also new metal chelating agents.<sup>20</sup>

In conclusion, a synthesis of new enantiopure 4-substituted 5-(*N*-Boc)aminopiperidin-2-ones of defined 4*S* and 5*R* configurations was developed taking advantage of stereospecific functionalization of the versatile bicyclic  $\alpha,\beta$ -unsaturated lactam **4**.

## Acknowledgements

We are grateful to Dr. Pierre Potier for a grant (O.C.).

## References

1. (a) Schneider, M. J. In *Alkaloids: Chemical and Biological Perspectives*; Pelletier, S. W. Ed.; Pergamon: Oxford, 1996; Vol. 10, pp. 155–299. (b) O'Hagan, D. *Nat. Prod. Reports* **1997**, *14*, 637–651.
2. Bailey, P. D.; Millwood, P. A.; Smith, P. D. *J. Chem. Soc., Chem. Commun.* **1998**, 633–640.
3. (a) Liu, K. K. C.; Kajimoto, T.; Chen, L.; Zhong, Z.; Ichikawa, Y.; Wong, C.-H. *J. Org. Chem.* **1991**, *56*, 6280–6289. (b) Kajimoto, T.; Liu, K. K. C.; Pederson, R. L.; Zhong, Z.; Ichikawa, Y.; Porco, J. A., Jr.; Wong, C.-H. *J. Am. Chem. Soc.* **1991**, *113*, 6187–6196. (c) Frankowski, A.; Seliga, C.; Bur, D.; Streith, J. *Helv. Chim. Acta* **1991**, *74*, 934–940. (d) Ganem, B. *Acc. Chem. Res.* **1996**, *29*, 340–347.
4. Asano, N.; Nash, R. J.; Molyneux, R. J.; Fleet, G. W. J. *Tetrahedron: Asymmetry* **2000**, *11*, 1645–1680.
5. Glänzer, B. I.; Györgydeák, Z.; Bernet, B.; Vasella, A. *Helv. Chim. Acta* **1991**, *74*, 343–369 and references cited therein.
6. (a) Kobayashi, J.; Ishibashi, M. *Heterocycles* **1996**, *42*, 943–970. (b) Ninomiya, I.; Kiguchi, T.; Naito, T. *Alkaloids* **1998**, *50*, 317–342.
7. Freyer, A. J.; Patil, A. D.; Killmer, L.; Troupe, N.; Mentzer, M.; Carte, B.; Faucette, L.; Johnson, R. K. *J. Nat. Prod.* **1997**, *60*, 986–990.
8. (a) Herdeis, C.; Aschenbrenner, A.; Kirfel, A.; Schwabenländer, F. *Tetrahedron: Asymmetry* **1997**, *8*, 2421–2432. (b) Panday, S. K.; Langlois, N. *Synth. Commun.* **1997**, *27*, 1373–1384 and references cited therein.
9. Langlois, N.; Calvez, O.; Radom, M.-O. *Tetrahedron Lett.* **1997**, *38*, 8037–8040.
10. Hughes, D. L. *Org. React.* **1992**, *42*, 335–656.
11. (a) Miller, M. J. *Acc. Chem. Res.* **1986**, *19*, 49–56. (b) Maurer, P. J.; Miller, M. J. *J. Org. Chem.* **1981**, *46*, 2835–2836.
12. Panday, S. K.; Langlois, N. *Tetrahedron Lett.* **1995**, *36*, 8205–8208.
13. Rotella, D. P. *Synlett* **1996**, 479–480.
14. Langlois, N.; Moro, A. *Eur. J. Org. Chem.* **1999**, 3483–3488.
15. Langlois, N. Second Euroconference on Marine Natural Products, Sept. 1999, Santiago de Compostela, Spain.
16. (a) Chen, Y.; Vogel, P. *J. Org. Chem.* **1994**, *59*, 2487–2496. (b) Veith, U.; Schwaradt, O.; Jäger, V. *Synlett* **1996**, 1181–1183.
17. (a) Keck, G. E.; McHardy, S. F.; Wager, T. T. *Tetrahedron Lett.* **1995**, *36*, 7419–7422. (b) Chiara, J. L.; Destabel, C.; Gallego, P.; Marco-Contelles, J. *J. Org. Chem.* **1996**, *61*, 359–360. (c) Falborg, L.; Jorgensen, K. A. *J. Chem. Soc., Perkin Trans. 1* **1996**, 2823–2826. (d) Keck, G. E.; Wager, T. T.; McHardy, S. F. *Tetrahedron* **1999**, *55*, 11755–11772.
18. This reagent is easier to handle than TiCl<sub>3</sub> generally used to reduce the corresponding hydroxamic acids.<sup>12,19</sup>
19. Mattingly, P. G.; Miller, M. J. *J. Org. Chem.* **1980**, *45*, 410–415.
20. Lucet, D.; Le Gall, T.; Mioskowski, C. *Angew. Chem. Int. Ed.* **1998**, *37*, 2580–2627.